Irreversibility on the Level of Single-Electron Tunneling

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OUTLINE

Second law of thermodynamic and Fluctuation theorem

Experimental setup (DQD)

Experimental demonstration of the Fluctuation theorem

close to equilibrium



far from equilibrium

Conclusions

Second law of thermodynamics and Fluctuation theorem

Kelvin statement There exists no thermodynamic transformation whose *sole* effect is to extract a quantity of heat from a given heat reservoir and to convert it entirely into work

Clausius Statement There exists no thermodynamic transformation whose *sole* effect is to extract a quantity of heat from a colder reservoir and to deliver it to a hotter reservoir

Clausius Theorem In any cyclic transformation throughout which the temperature is defined, the following inequality holds:

$$\oint \frac{dQ}{T} \le 0$$

Second law of thermodynamics and Fluctuation theorem

Entropy
$$S(A) := \int_0^A \frac{dQ}{T}$$
 integration over reversibel path joining 0
to A
 $S(B) - S(A) \ge \int_A^B \frac{dQ}{T}$

The entropy of a thermally isolated system never decreases

$$S(B) - S(A) \ge 0$$

The truth of the second law is ... a statistical, not a mathematical, truth, for it depends on the fact that the bodies we deal with consist of millions of molecules... Hence the second law of thermodynamics is continually being violated, and that to a considerable extent, in any sufficiently small group of molecules belonging to a real body. James Clerk Maxwell, 1878



Second law of thermodynamics and Fluctuation theorem

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Probability of Second Law Violations in Shearing Steady States

Denis J. Evans

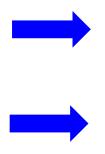
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G. P. Morriss School of Physics, University of South Wales, Kensington, New South Wales, Australia (Received 26 March 1993)

Classical fluid driven by an external shear

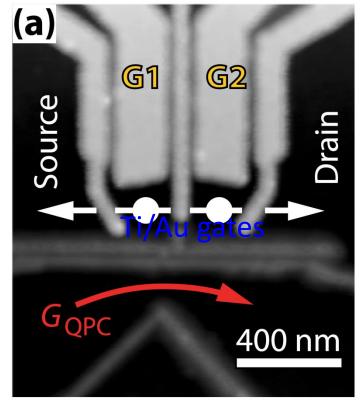
$$\frac{P_t(-\Delta S)}{P_t(\Delta S)} = \exp(-\Delta S) \qquad \Delta S = \frac{W}{T}$$



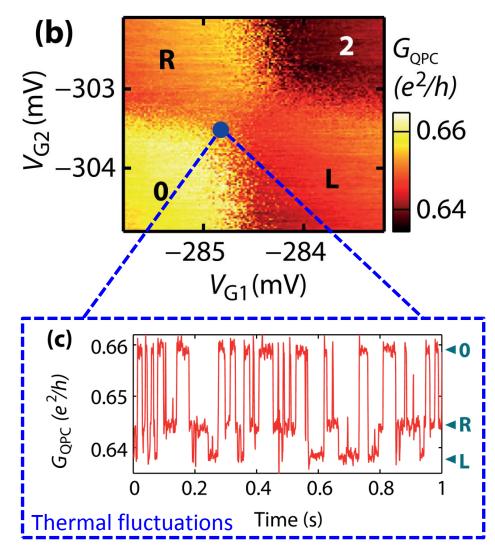
First experimental demonstration in 2002 in a classical experiment at room temperature [PRL **89**, 050601 (2002)].

This result has been anticipated in the quantum mechanical regime [Rev. Mod. Phys. 91, 1665 (2009)]

Experimental setup (double quantum-dot)



GaAs/Al_{0.3}Ga_{0.7}As heterostructure

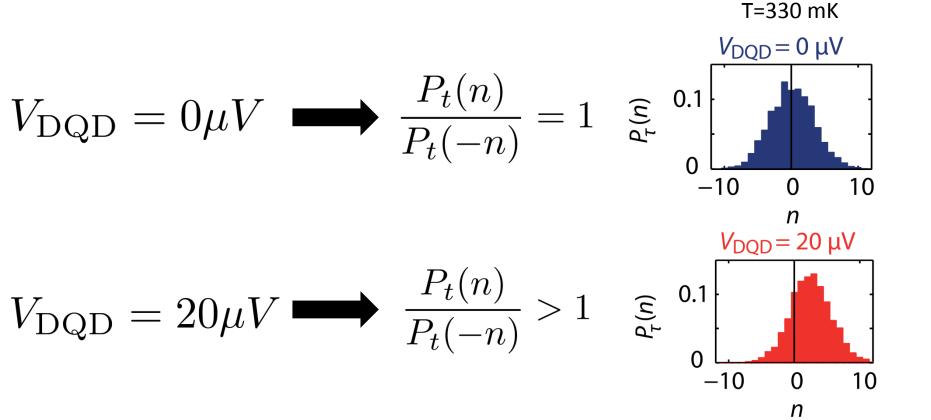


This experiment represents a step forward into this direction (still no quantum coherence here)

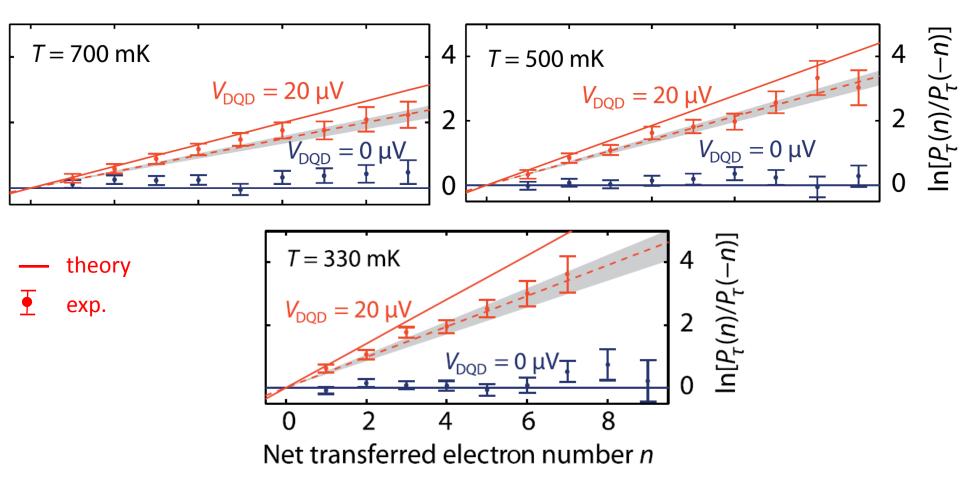
Experimental demonstration of the fluctuation theorem

$$\Delta S = neV_{\rm DQD}/T$$

$$\frac{P_t(n)}{P_t(-n)} = \exp\left(\frac{neV_{\rm DQD}}{k_BT}\right)$$

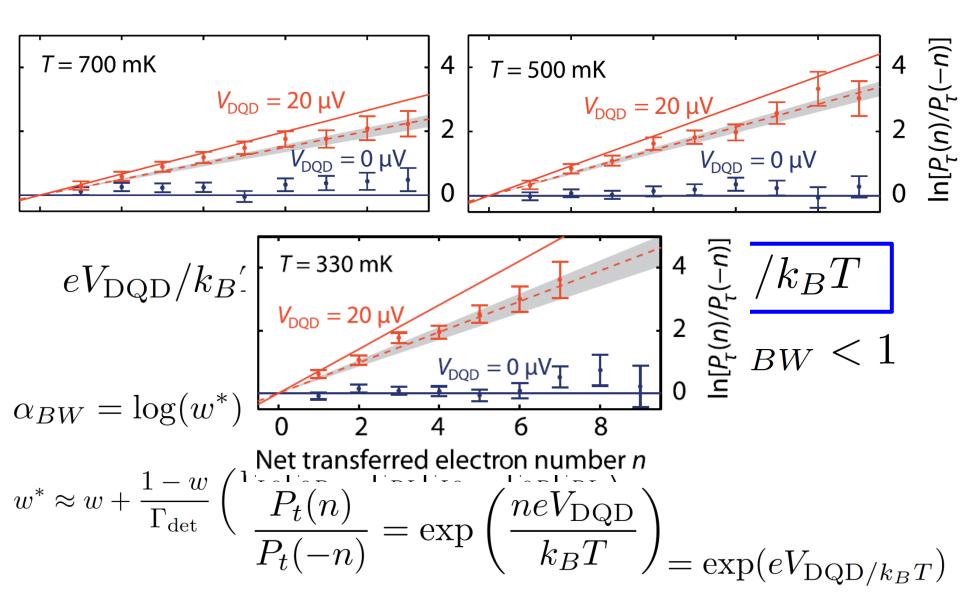


Experimental demonstration of the fluctuation theorem close to equilibrium



Where does the discrepancy between theory and experiment come from?

Experimental demonstration of the fluctuation theorem close to equilibrium

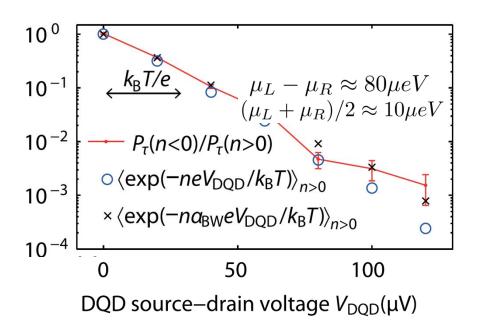


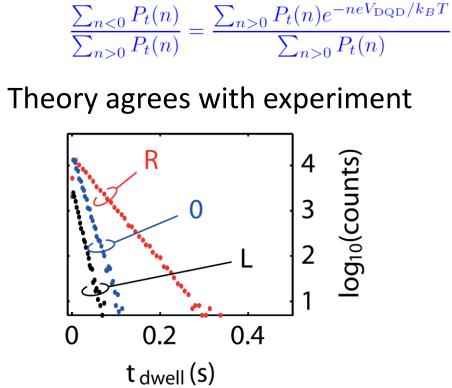
Experimental demonstration of the fluctuation theorem far from equilibrium

Until now $V_{DQD} = 20 \mu V$ and $k_B T/e$ between 28 and 58 μV

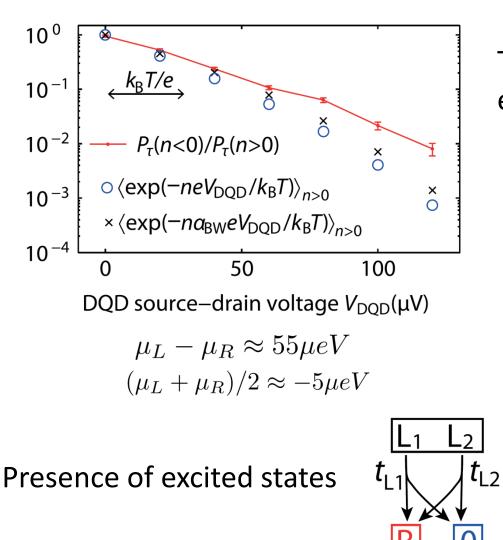
quite close to equilibrium

 V_{DQD} = 120 μ V at T=330 mK (far from equilibrium)

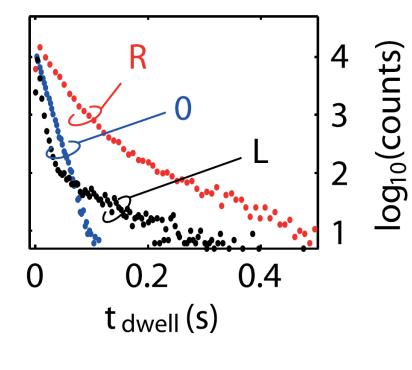




Experimental demonstration of the fluctuation theorem far from equilibrium



Theory does not agree with experiment



Conclusions

The second law of thermodynamic can be violated for small systems and small times



Experimental demonstration exist in the classical regime

Experimental demonstration in a double quantum-dot setup at small temperature



represents a step forward to the experimental demonstration in the quantum regime

THANK YOU !